LESSON PLAN

Discipline:	Semester:	Name of the Faculty: Mr Durga Prasanna Pattanaik
Electrical Engg	First (1)	
Subject:	No. of days/week	Semester from Date: 25.10.2022 to Date: 31.01.2023
Engg. Chemistry	class allotted: Six (6)	No. of Weeks: 15
WEEK	CLASS DAY	THEORY TOPICS
	st 1	PHYSICAL CHEMISTRY-
	_	Introduction on Atomic Structure, Fundamental Particles (Electron,
	nd	Proton & Neutron, Definition, Mass and Charge)
	2	Rutherford's Atomic model (Postulates and Failure)
st 1	rd 3	Atomic Mass & Mass Number
	th	Definition, Examples and Properties of Isotopes, Isobars, and Isotones
	4 th 5	Bohr's Atomic model (Postulates only)
	th 6	Bohr-Bury scheme
	st 1	Aufbau Principle
	nd 2	Hund's Rule, Electronic configuration (up to atomic no.30)
	rd 3	Review Class
nd 2	th	Definition and Types of Chemical Bond Definition of Electrovalent
_	4	bond with examples (NaCl , MgCl ₂)
	th 5	Definition of Covalent bond with examples (H ₂ , Cl ₂ , O ₂ , N ₂ , H ₂ O, CH ₄ ,
	th	NH ₃) Definition of Coordinate bond with examples (NH ₄ ⁺ , SO ₂)
	6 st	
	1	Review Class
	nd 2	Concept of Arrhenius theory for acid and base with examples
	rd	(Postulates and Limitations)
	3	Concept Lowry Bronsted theory for acid and base with examples (Postulates and Limitations)
rd		Concept Lewis theory for acid and base with examples
3	th	(Postulates and Limitations)
	4	Neutralization of Acid and Base.
	th	Definition of Salt, Types of salt (Normal, acidic, basic, double,
	5	complex, and mixed salts definitions with 2 examples from each)
	6 th	Review Class
	st 1	Monthly Test-01
th 4	nd 2	Definition of Atomic Weight, Molecular Weight & Equivalent weight
	rd 3	Determination of Equivalent Weight of Acid, Base, and Salt
	th 4	Modes of expression of Concentration (Molarity, Normality and
	th	Molality) Simple problems, pH of solution (Definition with Numericals)
	5 th	Importance of pH in industry(Sugar, textile, Paper industries)
	6 st	Review Class
	1 nd	
	2	Definition and types (Strong & Weak) of Electrolytes with Examples

	rd 3	Electrolysis (Principle & Process) with Example of NaCl (fused and
	th	aqueous solution)
	4	Faraday's 1 st law of Electrolysis.(Statement, Mathematical
th 5	th	expression)
	5	Simple problems
	th 6	Faraday's 2 nd law of Electrolysis.(Statement, Mathematical
	st	expression)
	1	Simple problems, Industrial application of Electrolysis-Electroplating (Zinc Only)
-	nd	Review Class
_	2 rd	
th	3	Definition of Corrosion, Types of Corrosion-
6	th	Atmospheric Corrosion, Waterline Corrosion Mechanism of Rusting of Iron Only
	4 th	- ,
	5	Protection from Corrosion by (i) Alloying and (ii) Galvanization
	th 6	Review Class
	st 1	Monthly Test-02
-	nd	INORGANIC CHEMISTRY –
	2	Definition of Mineral, ore, gangue, example. Distinction between Ores
th 7		and Mineral
,	rd 3	General methods of extraction of metal, (i) Ore Dressing
	th	(ii) Concentration. (Gravity Separation, Magnetic Separation)
	4	Froth floatation , Leaching
	th 5	iii) Oxidation (Calcinations, Roasting)
	th 6	Reduction (Smelting) Definition & examples of flux, slag
	st 1	Refining of ore (Electro refining & Distillation)
	nd 2	Review Class
	rd	Definition of alloy. Types of Alloys (Ferro, Non-Ferro,& Amalgam)
8 th	3	with Examples
8	th 4 th	Composition and uses of Brass, Bronze, Alnico, Duralumin
	tn 5	Review Class
	th 6	ORGANIC CHEMISTRY-
	_	Hydrocarbons: Saturated Hydrocarbons (Definition with examples)
	st 1	Hydrocarbons: Unsaturated Hydrocarbons
_	nd	(Definition with examples)
	2	Aliphatic and Aromatic Hydrocarbons (Huckel's Rule only).
	rd 3	Difference between Aliphatic and Aromatic hydrocarbons
9 th	th 4	IUPAC system of Nomenclature: Alkane, Alkene, Alkyne,
	4 th	IUPAC system of nomenclature of Alkyl halide and Alcohol (up to 6
	5	carbons) bond line notation.
	6 th	Use of some common aromatics compounds (Benzene, toluene, BHC, phenol, Naphthalene, Anthracene & Benzoic acid) in daily life.
	st 1	Review Class
	nd	Monthly Test-03
	2	

	rd	INDUSTRIAL CHEMISTRY –
10 th	3	Sources of water. Soft water, Hard water,
10	th 4	Types of Hardness (temporary or carbonate and permanent or non-
		carbonate)
	th 5	Removal of hardness by - Lime soda method
	th	(Hot lime : Principle, process & advantages)
	6"	Removal of hardness by - Lime soda method (Cold lime – Principle,
	st	process & advantages)
	1	Advantages of Hot lime over Cold lime process
	nd 2	Removal of hardness by -Organic Ion exchange method (Principle,
	rd	process, and regeneration of exhausted resins)
th	3	Review Class
11	th	Definition of lubricant, Types (Solid, liquid and Semisolid with
	4	examples only)
	th 5	Specific uses of lubricants (Graphite, Oils, Grease)
		Purpose of lubrication
	th 6	Review Class
	st	Definition and classification of fuel, Definition of calorific value of fuel,
	1	Choice of good fuel.
	nd 2	Liquid: Diesel, Petrol and Kerosene
		Composition and uses
12 th	rd 3	Gaseous: Producer gas and Water gas
		Composition and uses.
	th 4	Elementary idea about LPG , CNG and
_	th	Coal gas (Composition and uses only)
	5	Review Class
	th 6	Definition of Monomer, Polymer, Homopolymer, Co-polymer and
		Degree of polymerization
	st 1	Difference between Thermosetting and Thermoplastic
	nd	Composition and uses of Polythene & Poly-Vinyl Chloride and
	2	Bakelite
th	rd 3	Definition of Elastomer (Rubber), Natural Rubber (it's drawbacks),
13	th	Vulcanization of Rubber.
	4	Advantages of Vulcanized rubber over raw rubber
		3
	th 5	Review Class
	th	Monthly Test-04
	6 st	Pesticides, Insecticides, Herbicides, Fungicides. Examples and Uses
_	1 nd	
	2	Bio-fertilizers : Definition , Examples and Uses
th	rd 3	Review Class
14	th 4	Revision Class
	th	Revision Class
	5 th	Revision Class
	6 st	Revision Class
_	1 nd	
	2	Revision Class

15 th	rd 3	Revision Class
	th 4	Revision Class
	th 5	Revision Class
	6 th	Revision Class